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#### **Key indicators**

Single-crystal X-ray study T = 294 KMean  $\sigma(\text{C}-\text{C}) = 0.014 \text{ Å}$  R factor = 0.057 wR factor = 0.123 Data-to-parameter ratio = 17.5

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. In the title compound,  $[Pd(C_{12}H_{10}OP)Cl(C_{17}H_{15}N_2P)]$ , the Pd<sup>II</sup> atom adopts a square-planar coordination geometry.

Chloro[N-(diphenylphosphino)pyridin-2-amine-

 $\kappa^2 N^1$ , P](diphenylphosphoryl)palladium(II)

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## Comment

Phosphine ligands and their complexes play a very important role in coordination chemistry (Zhang & Cheng, 1996). Their synthesis and reactivity has attracted considerable interest owing to their novel structural and reactive features, and their potential application as catalysts (David *et al.*, 1997). Previously, we have reported an unexpected  $Pt^{II}$  complex with a rearranged phosphine ligand, *N*-(diphenylphosphino)-*N*-[(diphenylphosphoryl)methyl]pyridin-2-amine (Li *et al.*, 2006). Treatment of the above ligand with Pd(cod)Cl<sub>2</sub> (where cod is 1,5-cyclooctadiene) in dichloromethane yielded the unexpected title complex, (I), in which the phosphine ligand was cleaved and oxidized into two parts, which are coordinated to the Pd atom.



As shown in Fig.1, the Pd<sup>II</sup> atom is coordinated by one N and one P atom from an *N*-(diphenylphosphino)pyridin-2amine ligand, one P atom from a diphenylphosphoryl ligand and a chloride anion, in a square-planar coordination. The Pd-P bond *trans* to N is longer than that *trans* to Cl by *ca* 0.056 Å (Table 1). The dihedral angle between the fivemembered chelate ring and the pyridine ring is 6.4 (1)°.

## Experimental

The *N*-(diphenylphosphino)-*N*-[(diphenylphosphoryl)methyl]pyridin-2-amine ligand was prepared according to the literature method of Li *et al.* (2006). Complex (I) was obtained from the treatment of the above ligand (0.142 g, 0.25 mmol) with Pd(cod)Cl<sub>2</sub> (0.072 g, 0.25 mmol) in dichloromethane (yield 0.102 g, 66%). Single crystals

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# metal-organic papers

of (I) were obtained by slow evaporation of a dichloromethane solution.

Z = 8

 $D_r = 1.520 \text{ Mg m}^{-3}$ 

Mo  $K\alpha$  radiation

Block, pale yellow

 $0.22 \times 0.18 \times 0.16 \text{ mm}$ 

29378 measured reflections

5756 independent reflections

3130 reflections with  $I > 2\sigma(I)$ 

H atoms treated by a mixture of

 $w = 1/[\sigma^2({F_{\rm o}}^2) + 29.0746P]$ 

where  $P = (F_0^2 + 2F_c^2)/3$ 

independent and constrained

 $\mu = 0.93 \text{ mm}^{-1}$ 

T = 294 (2) K

 $R_{\rm int} = 0.099$ 

 $\theta_{\rm max} = 26.8^{\circ}$ 

refinement

 $(\Delta/\sigma)_{\rm max} = 0.001$ 

 $\Delta \rho_{\rm max} = 0.81 \text{ e} \text{ Å}^{-3}$ 

 $\Delta \rho_{\rm min} = -0.94 \text{ e} \text{ Å}^{-3}$ 

#### Crystal data

 $\begin{array}{l} \left[ Pd(C_{12}H_{10}OP)Cl(C_{17}H_{15}N_2P) \right] \\ M_r = 621.30 \\ Orthorhombic, Pbca \\ a = 13.226 \ (2) \ \text{\AA} \\ b = 13.594 \ (2) \ \text{\AA} \\ c = 30.205 \ (5) \ \text{\AA} \\ V = 5430.7 \ (15) \ \text{\AA}^3 \end{array}$ 

Data collection

Bruker SMART CCD area-detector diffractometer  $\varphi$  and  $\omega$  scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)  $T_{\min} = 0.822, T_{\max} = 0.866$ 

#### Refinement

Refinement on  $F^2$   $R[F^2 > 2\sigma(F^2)] = 0.057$   $wR(F^2) = 0.123$  S = 1.125756 reflections 329 parameters

## Table 1

Selected geometric parameters (Å, °).

Pd1-N1	2.162 (5)	Pd1-P2	2.2739 (18)
Pd1-P1	2.2178 (19)	Pd1-Cl1	2.3609 (19)
N1 - Pd1 - P1	82.61 (15)	N1-Pd1-Cl1	92.93 (16)
N1-Pd1-P2	174.06 (16)	P1-Pd1-Cl1	173.45 (7)
P1-Pd1-P2	91.48 (7)	P2-Pd1-Cl1	93.01 (7)

The N-bound H atom was located in a difference map and refined freely [N-H = 0.85 (6) Å]. The C-bound H atoms were positioned geometrically, with C-H = 0.93 Å and were constrained to ride on their parent atoms, with  $U_{iso}(H) = 1.2U_{ca}(C)$ .



#### Figure 1

The molecular structure of the title compound, showing 30% probability displacement ellipsoids and the atom-numbering scheme. H atoms have been omitted for clarity.

Data collection: *SMART* (Bruker, 1998); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1999); software used to prepare material for publication: *SHELXTL*.

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